

Imperial TC

JEE Advanced 2024 Detailed Syllabus, Pattern, and Marks Scheme

1. JEE Advanced Exam Pattern:

- Mode of Exam: Online (Computer-Based Test)
- Duration: 3 hours per paper (2 papers, Paper 1 and Paper 2)
- Sections: Three sections in each paper Physics, Chemistry, Mathematics
- Question Types: Multiple Choice Questions (MCQs), Numerical Value-based Questions, and Match the Following
- Total Questions: Varies each year; typically around 54 questions per paper
- Marks Scheme:
 - Varies each year, with a mix of full, partial, and zero marks.
 - No uniform negative marking scheme; it depends on the question type.
- Total Marks: Varies, usually around 360 marks (180 marks per paper)

2. Detailed Syllabus:

Physics:

1. General:

Dimensional analysis, experimental skills, units, and measurements.

2. Mechanics:

- Kinematics: Motion in one and two dimensions, relative velocity, projectiles.
- Newton's Laws: Applications to simple systems, friction, tension.
- Work, Energy, and Power: Conservation laws, work-energy theorem, potential energy.
- Systems of Particles: Center of mass, momentum conservation, collisions.
- Gravitation: Newton's law, potential energy, orbits, escape velocity.
- Rigid Body Dynamics: Rotational motion, moment of inertia, torque, angular momentum.
- o Fluid Mechanics: Pressure, Bernoulli's theorem, viscosity, surface tension.
- Thermal Physics: Kinetic theory, thermodynamics, laws of thermodynamics,
 Carnot engine.

3. Electricity and Magnetism:



- Electrostatics: Coulomb's law, electric field, Gauss's law, potential, capacitors.
- Current Electricity: Ohm's law, series and parallel circuits, Kirchoff's laws.
- Magnetism: Biot-Savart law, Ampere's law, Lorentz force, magnetic properties of materials.
- Electromagnetic Induction: Faraday's laws, Lenz's law, inductance, AC circuits.

4. Optics:

- o Geometrical Optics: Reflection, refraction, lenses, mirrors, optical instruments.
- Physical Optics: Interference, diffraction, polarization, wavefronts.

5. Modern Physics:

- Dual Nature of Matter and Radiation: Photoelectric effect, de Broglie waves.
- Atoms and Nuclei: Models of atoms, radioactivity, nuclear fission and fusion.
- Electronics: Semiconductors, diodes, transistors, logic gates, digital electronics.

Chemistry:

Physical Chemistry:

- 1. Basic Concepts in Chemistry: Atomic and molecular masses, mole concept, stoichiometry.
- 2. States of Matter: Gaseous state, ideal gas law, real gases, liquid state.
- 3. Atomic Structure: Bohr model, quantum mechanics, electronic configuration.
- 4. Chemical Bonding: Molecular orbital theory, VSEPR theory, hybridization, ionic and covalent bonds.
- 5. Chemical Thermodynamics: Enthalpy, entropy, free energy, Hess's law, spontaneity.
- 6. Solutions: Colligative properties, Raoult's law, ideal and non-ideal solutions.
- 7. Chemical Kinetics: Rate laws, reaction order, Arrhenius equation, collision theory.
- 8. Surface Chemistry: Adsorption, colloids, catalysis, emulsions.
- 9. Electrochemistry: Electrolytic cells, Nernst equation, electrochemical series, batteries.
- 10. Nuclear Chemistry: Radioactivity, decay processes, nuclear reactions.

Inorganic Chemistry:

1. Periodic Table: Trends in atomic and ionic sizes, ionization energy, electronegativity.



- 2. Chemical Bonding: Lewis structures, valence bond theory, crystal field theory.
- 3. Metallurgy: Extraction of metals, refining, and purification.
- 4. s-Block Elements: Properties of alkali and alkaline earth metals, important compounds.
- 5. p-Block Elements: Properties, trends, and compounds of groups 13-18.
- 6. d- and f-Block Elements: Properties of transition metals, lanthanides, and actinides.
- 7. Coordination Compounds: Nomenclature, bonding theories, isomerism, color, and magnetic properties.
- 8. Environmental Chemistry: Atmospheric pollution, ozone layer depletion, water and soil pollution.

Organic Chemistry:

- 1. General Organic Chemistry: Inductive, mesomeric, hyperconjugation effects, resonance, acidity, and basicity.
- 2. Hydrocarbons: Alkanes, alkenes, alkynes, aromatic hydrocarbons, reactions and mechanisms.
- 3. Haloalkanes and Haloarenes: Nucleophilic substitution reactions, SN1, SN2 mechanisms.
- 4. Alcohols, Phenols, and Ethers: Properties, acidity, reactions, and preparation methods.
- 5. Aldehydes, Ketones, and Carboxylic Acids: Reactivity, mechanisms, nucleophilic addition, oxidation-reduction reactions.
- 6. Amines: Structure, basicity, preparation, and reactions, diazotization.
- 7. Biomolecules: Carbohydrates, proteins, lipids, nucleic acids, their structure, and function.
- 8. Polymers: Classification, types, mechanisms, and uses.
- 9. Chemistry in Everyday Life: Drugs, food preservatives, detergents, and polymers.

Mathematics:

1. Algebra:

 Quadratic Equations: Roots, nature of roots, relationship between roots and coefficients.



- Sequences and Series: Arithmetic and geometric progressions, harmonic progression.
- Matrices and Determinants: Inverse, rank, properties, eigenvalues, and eigenvectors.
- o Complex Numbers: Argand plane, De Moivre's theorem, roots of complex numbers.
- o Probability: Bayes' theorem, probability distributions, random variables.
- Permutation and Combination: Factorials, binomial theorem, combinatorial identities.
- Binomial Theorem: General term, expansion, application in probability.

2. Trigonometry:

- o Trigonometric Functions: Identities, inverse trigonometric functions, graphs.
- Solution of Triangles: Laws of sines and cosines, area, and solutions.

3. Coordinate Geometry:

- Straight Lines: Equations, distance of a point, angle between lines.
- Circles: Tangents, chord properties, equations.
- Conic Sections: Parabolas, ellipses, hyperbolas, their properties and equations.

4. Calculus:

- Limits and Continuity: L'Hopital's rule, continuity, and differentiability.
- Differentiation: Derivatives, higher-order derivatives, applications in finding tangents, normals, maxima, and minima.
- Integration: Indefinite and definite integrals, properties, applications in area, and volume.
- Differential Equations: First-order differential equations, linear differential equations.

5. Vectors and 3D Geometry:

- Vectors: Operations, dot and cross product, applications in geometry.
- Three-Dimensional Geometry: Equations of planes and lines, distances, and angles.

6. Mathematical Reasoning:

Logic: Statements, truth tables, implications, and equivalences.



7. Statistics:

o Measures of Central Tendency: Mean, median, mode, variance, standard deviation.

3. Marks Distribution:

- Physics:
 - Paper 1 & Paper 2: Combined weightage with 18-20 questions, approximately 120 marks
- Chemistry:
 - Paper 1 & Paper 2: Combined weightage with 18-20 questions, approximately 120 marks
- Mathematics:
 - Paper 1 & Paper 2: Combined weightage with 18-20 questions, approximately 120 marks

Marking Scheme (Varies Year to Year):

- MCQs:
 - Full marks for correct answers.
 - Partial marks for partially correct responses.
 - Negative marks for incorrect answers.
- Numerical Value Questions:
 - Full marks for correct answers.
 - No negative marking.
- Match the Following:
 - Full and partial marks depending on correctness.

P.S Please cross-check the syllabus on the official JEE website to ensure accuracy and stay updated with any changes.

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